

Short Communication

Ethylene rather than acetylene inhibits soil methane oxidation rates in a subtropical evergreen forest

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ABSTRACT

Ethylene and acetylene have been widely applied as inhibitors for microbial processes such as denitrification, nitrification, methanogenesis, and methane oxidation. Here, we tested the inhibition effects of these two trace gases on methane oxidation rates in subtropical evergreen forest soils. One-week laboratory incubations showed that aerobic soil methane oxidation rates at ambient or 50 parts per million (ppm) methane concentration were not affected by the addition of 100 or 100,000 ppm acetylene. In contrast, increasing amounts of ethylene markedly reduced soil methane oxidation rates. Our results suggest the selective inhibition of soil methane oxidation rates between ethylene and acetylene in these subtropical evergreen forest soils, and highlight the importance of verifying the efficiency of commonly-applied inhibitors before investigating microbial processes in underrepresented environments.

Atmospheric methane (CH₄) is a key driver of global warming with a 34 × higher warming potential than carbon dioxide (CO₂) (IPCC, 2013). Atmospheric CH₄ concentrations are a result of production and oxidation processes in various environments such as wetlands, rice fields, landfills, and forest soils (Aronson et al., 2013). In particular, forest soils have been identified as contributing significantly to an estimated 2.5–7.5% global CH₄ sink (Kolb, 2009; Aronson et al., 2013).

Aerobic CH₄ oxidation is mediated by CH₄-oxidizing bacteria (methanotrophs) that use CH₄ as their main carbon and energy source. They are found within the Verrucomicrobia, NC10, Gammaproteobacteria (Type I methanotrophs), and Alphaproteobacteria (Type II methanotrophs) (Knief, 2015). All aerobic methanotrophs oxidize CH₄ via methanol and formaldehyde to CO₂ (Hanson and Hanson, 1996; Bowman, 2006; McDonald et al., 2008; Trotsenko and Murrell, 2008). The first step in this pathway is catalyzed by methane monooxygenases (MMO) (McDonald et al., 2008; Knief, 2015).

In forest soils, methanotrophic activity has been linked to high-affinity methanotrophs that can grow at low atmospheric CH₄ concentrations (Bender and Conrad, 1992; Pratscher et al., 2018). Recently Pratscher et al. (2018) used fluorescence-labeled acetylene (C₂H₂) to demonstrate a link between the high-affinity activity of CH₄ oxidation

and specific methanotrophic cells *in situ*. It is well known that both C₂H₂ and ethylene (C₂H₄) have been commonly used to inhibit soil CH₄ oxidation (Prior and Dalton, 1985; Bender and Conrad, 1992; Chan and Parkin, 2000; Jäkel et al., 2004; Crombie and Murrell, 2014; Pratscher et al., 2018). However, few studies have focused on the effect of these two inhibitors on soil CH₄ oxidation in forest ecosystems (Chan and Parkin, 2000). Here, we selected soil from an evergreen subtropical forest to test the inhibition potential of C₂H₂ and C₂H₄ on CH₄ oxidation. We hypothesized that C₂H₂ has stronger inhibition effects on soil CH₄ oxidation than C₂H₄, as the molecular weight of the former is smaller and more similar to CH₄, the primary substrate of MMO.

The soil was collected from a subtropical evergreen forest in Tiantong National Forest Park, Eastern China (29°52'N, 121°39'E, 200 m above sea level). The region is dominated by a typical subtropical monsoon climate, with a mean annual temperature and precipitation of 16.2 °C and 1374.7 mm, respectively (Bu et al., 2018). We established three treatments in July 2013 (Fig. S1) (Bu et al., 2018): an ambient treatment (control), a 70% rainfall reduction treatment (drought), and a shade treatment (disturbance) to investigate the effects of drought on soil CH₄ oxidation. Soil samples from the different treatments were collected in August 2016, December 2016, and February 2017 and incubated in 1-L sealed flasks with 0, 100, or

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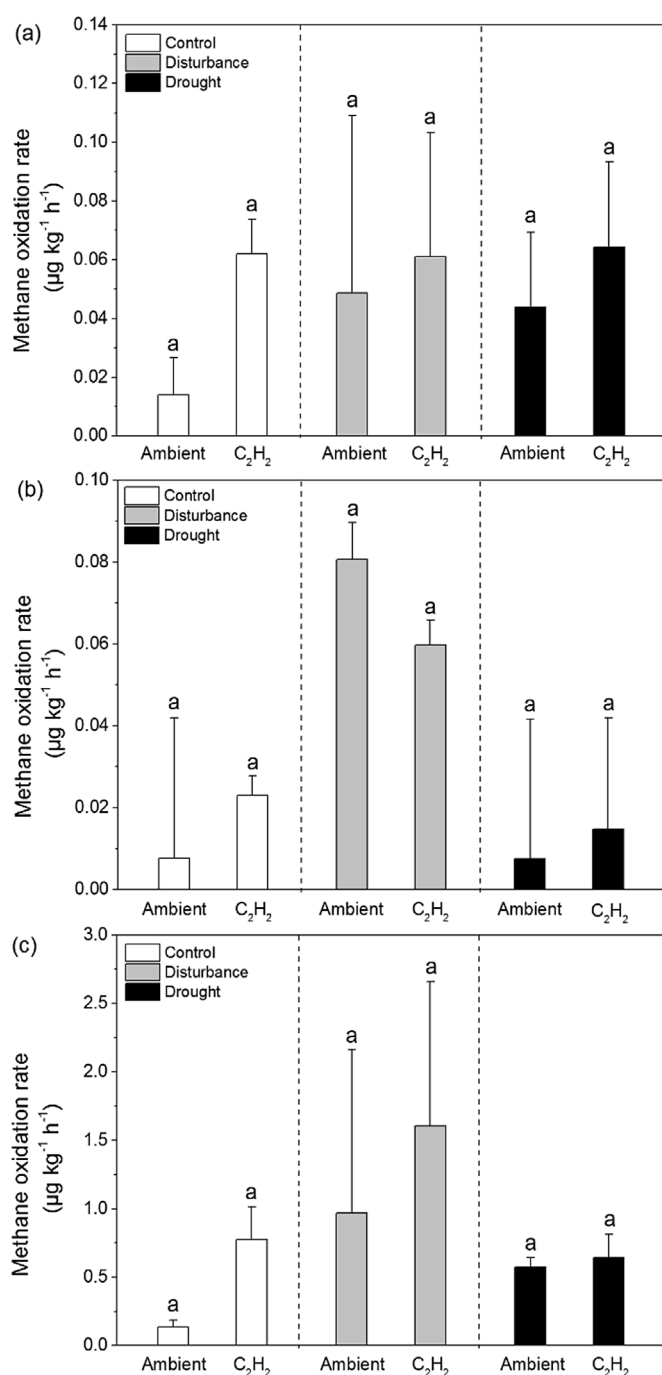


Fig. 1. Variations in the methane oxidation rate in the ambient atmosphere with or without 100 ppm C₂H₂ after 7 days of incubation in soils collected in August 2016 (a), December 2016 (b), and February 2017 (c) from a subtropical evergreen forest. Positive values indicate the soil methane oxidation rate ($n = 3$).

100,000 ppm C₂H₂ and ambient air in the headspace (containing ~2 ppm CH₄) or 50 ppm CH₄ in the dark at 22 °C for 5 h or 7 days. In a second experiment, we incubated soil samples from the control sites collected in February 2017 with 3, 5, 10, 15, 20, 30, and 50 ppm C₂H₄ and ambient air in the headspace at 22 °C for 7 days to compare the effect of C₂H₄ on soil CH₄ oxidation rates. To further compare the effect of C₂H₄ and C₂H₂ on soil CH₄ oxidation rates, we also tested soil samples collected in February 2017 with 5 ppm C₂H₄ and 100 ppm C₂H₂ and ambient air in the headspace at 22 °C in 1-day incubations. Microbial community compositions were measured via 16S rRNA high-

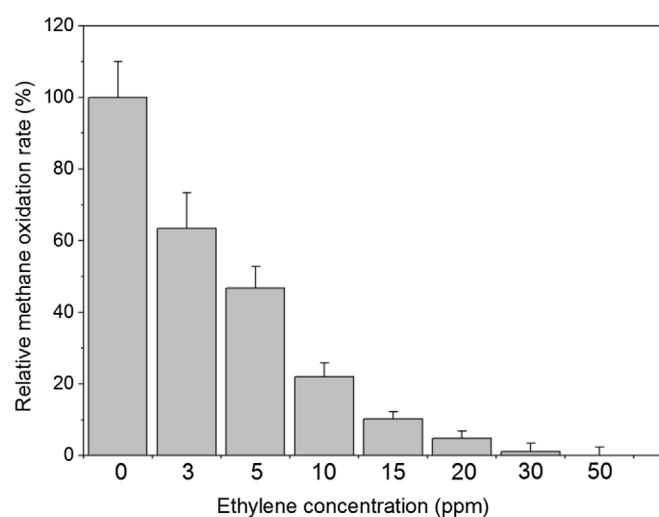


Fig. 2. Inhibition of methane oxidation by increasing the concentration of ethylene from 0 ppm to 50 ppm in soils of the control plots collected in February 2017 from a subtropical evergreen forest ($n = 3$).

throughput sequencing from the soil samples collected in August and December 2016. More detailed information about the experimental site and the process of data analysis are given in the Supplementary Materials.

We observed large variations in soil CH₄ oxidation rates at different sampling times during the 7-day incubations (Fig. 1). In addition, CH₄ oxidation rates were highest (approximately 10-fold) in soils collected in February 2017 (Fig. 1c). Irrespective of these changes, no significant differences in CH₄ oxidation rates were detected across all soil samples (treatments and sampling times) with 100 ppm C₂H₂ (Fig. 1). To verify these initial findings we performed additional incubations with samples from February 2017. Although the 5-h and 7-day incubations with 100,000 ppm C₂H₂ and 50 ppm CH₄ depicted similar results (Fig. S2), we found that the addition of 100 ppm C₂H₂ significantly decreased the soil CH₄ oxidation rates by 59% only at the control site in the 1-day incubations (Fig. S3).

In contrast, the addition of C₂H₄ had a remarkable inhibitory effect on soil methanotrophic activities in the 7-day incubations (Fig. 2). With 3 ppm C₂H₄ in the headspace, the soil CH₄ oxidation rate was inhibited by nearly 40%; at 10 ppm, it was inhibited by nearly 80%; it was completely inhibited at 50 ppm C₂H₄ (Fig. 2). We also observed that the addition of 5 ppm C₂H₄ produced consistent inhibition trends on soil CH₄ oxidation rates across the treatments in the 1-day incubations (Fig. S3) but with significantly lower soil CH₄ oxidation rates at the control site only (Fig. S3).

Many studies have shown that C₂H₂ can inhibit soil CH₄ oxidation rates (Prior and Dalton, 1985; Bender and Conrad, 1992; Crombie and Murrell, 2014; Pratscher et al., 2018). For instance, Prior and Dalton (1985) reported that C₂H₂ inhibited CH₄-oxidizing activity because C₂H₂ binds to the MMO at the active site. Chan and Parkin (2000) also showed that C₂H₂ had a strong inhibitory effect on the first-order rate constants of CH₄ oxidation and that 0.01% (100 ppm) C₂H₂ can completely inhibit CH₄ oxidation. We noticed that most of the previous studies examined the inhibition effects of C₂H₂ on soil CH₄ oxidation rates in 30-h incubations. Our results on the inhibition effects of soil CH₄ oxidation rates in 1-day incubations partially support these previous results, but we only observed significant decreases in soil CH₄ oxidation rates with the addition of C₂H₂ at the control site (Fig. S3). Our results indicate that the inhibition of CH₄ oxidation follows different dynamics in these soils than in those used in previous studies.

We did not find any evidence that C₂H₂ was toxic to cells because no differences in soil microbial activity were seen across the treatments and sampling times in the 1-week incubations (Fig. S4). Another reason

for the observed responses might be attributed to the binding stability between C_2H_2 and MMO. We suspect that C_2H_2 may initially have bound closely to the active site of MMO, which then resulted in significant decreases in soil CH_4 oxidation rates at the control sites in the 1-day incubations. However, the binding between C_2H_2 and MMO might become unstable over time, resulting in no marked inhibition effects of C_2H_2 on soil CH_4 oxidation rates in the 7-day incubations (Fig. 1). We did not find any other study reporting that C_2H_2 does not inhibit soil CH_4 oxidation. However, Schmidt and colleagues (2001) showed that ammonia monooxygenase (AMO), a MMO homolog, significantly reduced the C_2H_2 inhibition of ammonia oxidation activity for cells incubated in the presence of nitric oxide. They suggested that nitric oxide and C_2H_2 compete for the same binding site on the AMO. Hence, an unknown mechanism for MMO may be responsible for the observed results.

The soil microbial community that contributes to methane consumption was dominated by members of the genus *Methylobacter* (a Type I methanotroph) and the genus *Methylocella* (a Type II methanotroph) (Fig. S5). In particular, the genus *Methylocella* and its metabolic flexibility may provide a mechanism that explains the reduced inhibition potential of C_2H_2 in this subtropical evergreen forest soil. First, these organisms possess only the so-called soluble MMO (sMMO), which have been shown to be able to degrade other compounds or pollutants than the more widely distributed particulate MMO (Dedysh et al., 2005; Semrau, 2011). More importantly, Crombie and Murrell (2014) identified an additional soluble di-iron center monooxygenase (SDIMO) in the strain *Methylocella silvestris* that is involved in short-chain alkane (i.e. propane) metabolism. They demonstrated that growth on propane was completely inhibited in the presence of 2% C_2H_2 . These results also suggest that the representatives of the genus *Methylocella* found in this study may contain this additional SDIMO. As a result, this enzyme may have bound C_2H_2 in our experiments, thereby allowing the sMMO to keep oxidizing CH_4 . However, this mechanism still remains speculative and needs to be verified in further studies.

The observed inhibition of soil CH_4 oxidation rates by C_2H_4 in the 7-day incubations was in line with previous studies. For instance, Jäkel et al. (2004) showed that soil CH_4 uptake was nearly completely inhibited at a concentration of 10 ppm C_2H_4 in deciduous forest soil. Our previous studies confirmed that C_2H_4 acted as an inhibitor of CH_4 oxidation *in situ*, as this level of C_2H_4 concentration can be easily reached following a plant stress event (Zhou et al., 2013, 2018). We noticed that addition of 5 ppm C_2H_4 did not significantly inhibit soil CH_4 oxidation rates at the disturbance and drought sites. We observed higher *in situ* C_2H_4 concentrations at the drought sites (2.04 ± 0.75 ppm) than at the control sites (0.22 ± 0.09 ppm) (data not shown), which might suggest the resistance of soil CH_4 oxidation. We did not observe significant differences in *in situ* C_2H_4 concentrations between the disturbance and the ambient control sites, which suggests that short-term inhibition effects might occur.

In conclusion, our study showed that C_2H_2 , a commonly applied inhibitor for metalloenzymes such as nitrogenase, AMO, and MMO (Hyman and Wood, 1985) did not always inhibit CH_4 oxidation rates in a subtropical evergreen forest soil. This was unexpected but highlights the importance of verifying common research methodologies in novel and underrepresented environments. It further suggests that the metabolic potential of specific methanotrophs (e.g. *Methylocella*) may be underestimated in environments where CH_4 and other plant-derived hydrocarbons may co-occur.

Declarations of interest

The authors declare no conflict of interest.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.soilbio.2019.04.001>.

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